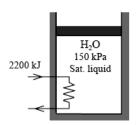
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Quiz 3

1. An insulated piston-cylinder device contains 0.005 m^3 of saturated liquid water at a constant pressure of 150 kPa. An electric resistance heater inside the cylinder is now turned on, and 2200 kJ of energy is transferred to the steam. Determine the entropy change of the water during this process.



Saturated water-Pressure table

		Specific volume, m³/kg		<i>Internal energy,</i> kJ/kg			Enthalpy, kJ/kg			Entropy, kJ/kg + K		
Press., <i>P</i> kPa	Sat. temp., 7 _{sat} °C	Sat. liquid, Vy	Sat. vapor, v _g	Sat. liquid, <i>u_f</i>	Evap., <i>u_{lg}</i>	Sat. vapor, <i>u_g</i>	Sat. liquid, h _f	Evap., h _g	Sat. vapor, <i>h_g</i>	Sat. liquid, s _r	Evap., s _é	Sat. vapor, s _g
75 100	91.76 99.61	0.001037 0.001043	2.2172 1.6941	384.36 417.40	2111.8 2088.2	2496.1 2505.6	384.44 417.51	2278.0 2257.5	2675.0	1.2132 1.3028		7.3589
101.325 125 150	99.97 105.97 111.35	0.001043 0.001048 0.001053	1.6734 1.3750 1.1594	418.95 444.23 466.97	2087.0 2068.8 2052.3	2506.0 2513.0 2519.2	419.06 444.36 467.13	2256.5 2240.6 2226.0	2675.6 2684.9 2693.1	1.3069 1.3741 1.4337	6.0476 5.9100 5.7894	7.2841

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2. An insulated piston-cylinder device contains 0.05 m^3 of saturated refrigerant 134a vapor at 0.8 MPa pressure. The refrigerant is now allowed to expand in a reversible manner until the pressure drops to 0.4 MPa. Determine (a) the final temperature in the cylinder and (b) the work done by the refrigerant.

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R-134a 0.05 m ³ 0.8 MPa	

Т	v	u	h	s						
°C	m³/kg	kJ/kg	kJ/kg	kJ/kg · K						
	$P = 0.80 \text{ MPa} (T_{\text{sat}} = 31.31 \text{°C})$									
Sat.	0.025621	246.79	267.29	0.9183						
40	0.027035	254.82	276.45	0.9480						
50	0.028547	263.86	286.69	0.9802						
60	0.029973	272.83	296.81	1.0110						
70	0.031340	281.81	306.88	1.0408						
80	0.032659	290.84	316.97	1.0698						
90	0.033941	299.95	327.10	1.0981						
100	0.035193	309.15	337.30	1.1258						
110	0.036420	318.45	347.59	1.1530						
120	0.037625	327.87	357.97	1.1798						
130	0.038813	337.40	368.45	1.2061						
140	0.039985	347.06	379.05	1.2321						
150	0.041143	356.85	389.76	1.2577						
160	0.042290	366.76	400.59	1.2830						
170	0.043427	376.81	411.55	1.3080						
180	0.044554	386.99	422.64	1.3327						

Saturated refrigerant-134a—Pressure table

		ic volume, ^{13/} kg	Inte	Internal energy, kJ/kg			Enthalpy, kJ/kg			Entropy, kJ/kg · K		
Press. P kPa	· · · · · ·		Sat. vapor, ^v g	Sat. liquid, <i>u</i> r	Evap., <i>u_{fg}</i>	Sat. vapor, ^u g	Sat. liquid, h _f	Evap., <i>h_{fg}</i>	Sat. vapor, <i>h_g</i>	Sat. Iiquid, <i>s</i> r	Evap., ^S fg	Sat. vapor, <i>s_g</i>
240	-5.38	0.0007620	0.083897	44.48	182.67	227.14	44.66	202.62	247.28	0.17794	0.75664	0.93458
280	-1.25	0.0007699	0.072352	49.97	179.50	229.46	50.18	199.54	249.72	0.19829	0.73381	0.93210
320	2.46	0.0007772	0.063604	54.92	176.61	231.52	55.16	196.71	251.88	0.21637	0.71369	0.93006
360	5.82	0.0007841	0.056738	59.44	173.94	233.38	59.72	194.08	253.81	0.23270	0.69566	0.92836
400	8.91	0.0007907	0.051201	63.62	171.45	235.07	63.94	191.62	255.55	0.24761	0.67929	0.92691